

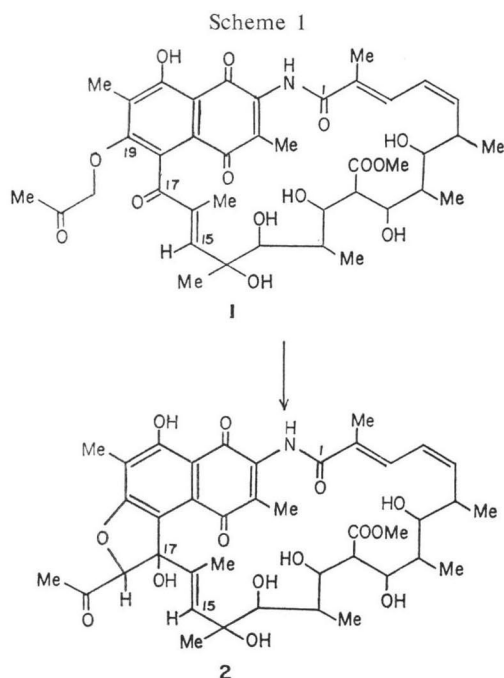
CHEMICAL MODIFICATION OF
STREPTOVARICIN C. II
THE INTRAMOLECULAR ALDOL
CONDENSATION OF 19-O-ACETONYL-
DAMAVARICIN C AND
ITS ANALOGS

Sir:

In the previous paper¹⁾ we reported a series of 19-O-substituted derivatives of damavaricin C, a degradative derivative of streptovaricin C, along with their biological properties. We wish to describe here the intramolecular aldol condensation of 19-O-acetyldamavaricin C¹⁾ and its analogs which gives products with a new carbon skeleton structure still possessing biological activity against gram-positive bacteria including *Mycobacterium tuberculosis in vitro*.

Upon treatment with benzylamine in benzene at room temperature, 19-O-acetyldamavaricin C (1) was converted into the isomer (2) [C₄₀H₅₁NO₁₄·H₂O; * amorphous orange powder, m.p. 162~163°C; λ_{max}^{MeOH} 270(29,600), 333(9,470) and 422(4,700) nm] by an intramolecular aldol condensation without dehydration (Scheme 1).

The n.m.r. spectra of 1 and the aldol compound 2 showed that the typical AB pattern



* Microanalysis agree with the molecular formula assigned.

of the acetyl CH₂ protons at δ4.37 (d, J=16.4 Hz) and δ4.84 (d, J=16.4 Hz) in 1 was replaced in 2 by two sharp singlets observed at δ4.84 and δ5.65. The latter disappeared by adding D₂O, indicating that compound 2 is an aldol form of 1. Furthermore, the vinyl proton at C-15, observed as a broad singlet at δ6.07 in 1, was shifted to higher field at δ4.96 in 2, indicating that the carbonyl group at C-17 in 1 was absent in 2 due to the intramolecular aldol condensation.

In the mass spectrum of 2, the molecular ion peak was not detected but dehydration peaks at *m/e* 751 (M-H₂O), 733 (M-2H₂O), 715 (M-3H₂O) and 697 (M-4H₂O) suggested

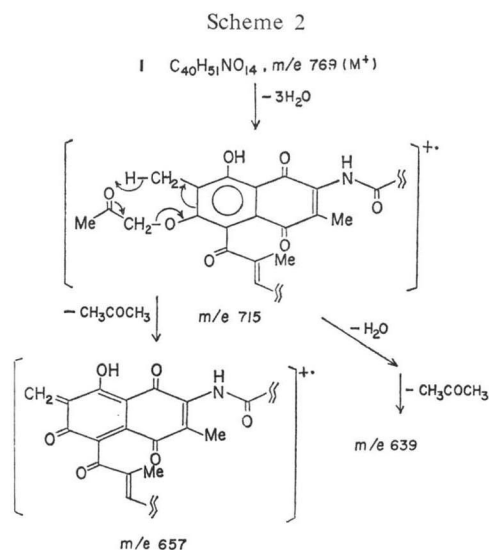
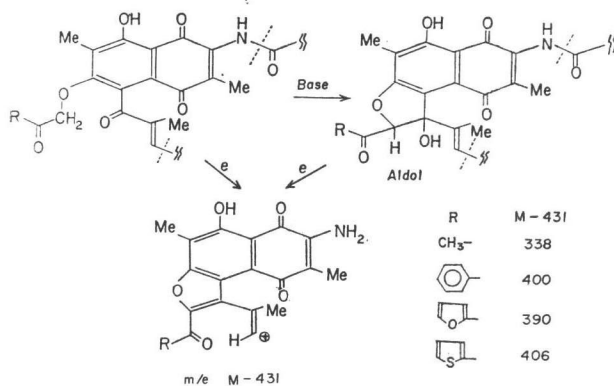


Table 1. Antibacterial activity (μg/ml) of various aldol compounds *in vitro*.

No.	R*	<i>St. aureus</i> FDA209P	<i>M. tuberculosis</i> H ₃₇ R _V	<i>E. coli</i>
2	CH ₃ -	0.2	50	>100
3	CH ₃ CH ₂ -	10	10	>100
4		2	50	>100
5		10	10	>100
6		5	10	>100
7		2	20	>100
8		5	10	>100

* R refers to the structure shown in Scheme 3.

Scheme 3



the facile loss of a tertiary hydroxyl group at the C-17 position in **2**. Furthermore, intense peaks due to deacetonation observed at m/e 657 and 639 in **1** were undetectable in the aldol compound **2**. The deacetonation may result from a vinylogous MCLAFFERTY rearrangement, as shown in Scheme 2.

Other than the aldol compound **2**, a series of analogous aldol compounds was prepared by the same reaction procedure.

Table 1 shows that various aldol compounds have an inhibitory activity against *Staphylococcus aureus* FDA209P and *Mycobacterium tuberculosis* H37Rv *in vitro*, but are inactive below 100 $\mu\text{g}/\text{ml}$ against *Escherichia coli* *in vitro*.

Preliminary investigations of other biological activities such as antitumor activity *in vitro* and the inhibition of RSV RNA-directed DNA

polymerase (reverse transcriptase) are now in progress and will be published elsewhere.²⁾

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